

#### **Cambridge Assessment International Education**

Cambridge International General Certificate of Secondary Education

COMBINED SCIENCE 0653/61

Paper 6 Alternative to Practical

October/November 2017

MARK SCHEME
Maximum Mark: 60

#### **Published**

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#### Cambridge IGCSE – Mark Scheme **PUBLISHED** 0653/61

Question	Answer	Marks
1(a)	Syringe / burette ;	1
1(b)	3 <b>and</b> 2 ;	1
1(c)	Axes labelled time <b>and</b> s (on y) concentration <b>and</b> % (on x) ;	4
	Suitable linear scale using at least half the grid ;	
	all 4 points plotted correctly $\pm$ half small square ;	
	Best fit line ;	
1(d)	Decreasing concentration increases time ORA;	1
1(e)(i)	All temps below 100;	2
	At least 3 below 50 must be above 0 ;	
1(e)(ii)	two from: Volume of milk Same type of milk pH concentration of enzyme volume of enzyme volume of water	1

© UCLES 2017 Page 2 of 7

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Question	Answer	Marks
2(a)(i)	T <sub>1</sub> 21.5; T <sub>2</sub> 34. <u>0</u> ;	2
2(a)(ii)	(+) 12.5;	1
2(a)(iii)	exothermic;	1
2(b)(i)	alkaline ;	1
2(b)(ii)	limewater / calcium hydroxide / Ca(OH) <sub>2</sub> ;	1
2(b)(iii)	sodium hydroxide ;	1
2(c)	(H is) calcium (oxide);	2
	<b>H</b> + water gives limewater for CO <sub>2</sub> test / calcium oxide reacts exothermically with water / <b>F</b> must be calcium hydroxide / <b>F</b> is limewater ;	
2(d)	chloride / Ct ;	1

0653/61

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Question	Answer	Marks
3(a)	1.4 ; 0.32 ;	2
3(b)(i)	to prevent wire getting hot / resistance of wire changing / cell running down;	1
3(b)(ii)	ammeter shows a reading ;	1
3(c)(i)	W/watt(s);	1
3(c)(ii)	0.23 <b>and</b> 0.17;	2
	2 decimal places ;	
3(d)(i)	a straight line with a positive gradient ;	2
	through the origin;	
3(d)(ii)	(actual values used to show that (for example)), doubling $\it I$ does not double $\it P$ ;	1

© UCLES 2017 Page 4 of 7 0653/61

# Cambridge IGCSE – Mark Scheme **PUBLISHED**

#### October/November 2017

Question	Answer	Marks
4(a)	Root hair ;	1
4(b)(i)	Quality drawing using at least half the box ;	3
	Nucleus correctly labelled ;	
	Cell wall correctly labelled ;	
4(b)(ii)	34 ± 1 ;	1
4(b)(iii)	Measurement to nearest mm;	1
4(b)(iv)	Magnification correctly calculated ;	1
4(c)	Starch present;	1
4(d)	select anther;	2
	use a microscope to observe ;	

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# Cambridge IGCSE – Mark Scheme **PUBLISHED**

Question	Answer	Marks
5(a)(i)	6.96 ;	2
	6.85 ;	
5(a)(ii)	0.49 and 0.49 ;	2
	+ and – ;	
5(b)	any <b>two</b> from: stays as blue ;	2
	mass changes are the same at the electrodes ;	
	anode dissolves / copper ions from anode go into solution ;	
5(c)(i)	(iron and copper) because copper dissolves from the anode / positive / plates on the cathode / negative ;	1
5(c)(ii)	copper sulfate (solution) ;	1
5(c)(iii)	smaller	2
	pink / orange / copper coloured	
	blue / unchanged ;;	
	all 3 correct = 2 marks; 1 or 2 correct = 1 mark	

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# Cambridge IGCSE – Mark Scheme **PUBLISHED**

Question	Answer	Marks
6(a)(i)	correct position marked ;	1
6(a)(ii)	0.87;	1
6(a)(iii)	data to 2 sf / <u>large</u> variation in raw data ;	1
6(b)(i)	0.76, 0.98, 1.21 correct answers only ;	1
6(b)(ii)	plots correct to half a small square – at least 4 correct ;	2
	good best fit line judgement ;	
6(b)(iii)	indication on graph of how data obtained AND at least half of line used;	2
	correct calculation for triangle method using data from graph ;	
6(c)(i)	correct answer from candidate's gradient value ;	1
6(c)(ii)	reduces percentage error in the time / reduces the effect of (human) reaction error;	1

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